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Reg. No. :

Name :

First Semester M.Sc. Degree Examination, December 2019

Chemistry/Polymer Chemistry

CH/CL/CM/CA/PC 213-PHYSICAL CHEMISTRY I

(Common for Chemistry (2016 Admission Onwards)) and

Polymer Chemistry (2018 Admission Onwards)

Time : 3 Hours

Max. Marks: 75

SECTION – A

Answer any **two** from a,b,c, of each question. Each question carries **2** marks.

- 1. (a) Calculate de Broglie wave length of mass 0.1gm moving with a velocity of 10 ms^{-1} .
 - (b) Explain uncertainty principle and its consequences.
 - (c) Define Hermitian operator.
- 2. (a) Discuss any one application of third law of thermodynamics.
 - (b) Explain the term fugacity.
 - (c) What are state functions? Explain with example.
- 3. (a) Explain the cleansing action of soap.
 - (b) Write any two examples for homogeneous catalysis.
 - (c) Give a brief account on enzyme catalysis.

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- 4. (a) Calculate the temperature at which the RMS velocity of nitrogen equals that of CO₂.
 - (b) Discuss briefly the effect of temperature and pressure on transport properties of gases.
 - (c) Give Radhich-Kwong equation. What are the limitations of this equation?
- 5. (a) Explain Taft's equation.
 - (b) What are the factors that affect rate of a chemical reaction?
 - (c) Define quantum yield.

(10 × 2 = 20 Marks)

SECTION – B

Answer either (a) or (b) of each question. Each question carries 5 marks.

- 6. (a) Distinguish between parallel and opposing reactions explain with suitable example.
 - (b) Write the decomposition kinetics of acetaldehyde.
- 7. (a) Explain two methods for the determination of surface tension of liquids.
 - (b) Explain the Virial equation of state and explain the terms.
- 8. (a) What is standard free energy of formation (ΔG) of Give its derivation.
 - (b) Explain Clausius equation with derivation.
- 9. (a) Write any two methods for the determination of surface area.
 - (b) Write the principle of photoelectron spectroscopy.
- 10. (a) Prove that eigen values of Hermitian operator are real.
 - (b) Derive time dependent schrodinger equation.

(5 × 5 = 25 Marks)



SECTION - C

Answer any **three** questions. Each question carries **10** marks

- 11. Explain fast reaction and describe any two methods to study fast reaction kinetics.
- 12. Discuss quantum mechanical aspect for a free particle in one dimensional box, What are application of Hermite polynomials. (7+3)
- 13. Derive Gibb's Helmholtz equation and its applications.
- 14. (a) Illustrate collision theory of reaction rates. What are the limitations of this theory?
 - (b) Discuss RRKM theory.

(5+5)

15. Discuss Maxwell's distribution of molecular velocities. Explain the various Maxwell's types of velocities.

(3 × 10 = 30 Marks)